

Types of Radioactive Decay

type	example	notes
alpha (α) decay	${}_{92}^{238}\text{U} \rightarrow {}_2^4\text{He} + {}_{90}^{234}\text{Th} + 2 {}_0^0\gamma$	<ul style="list-style-type: none"> ● α particle = ${}_2^4\text{He}$ nucleus (i.e., ${}_2^4\text{He}^{2+}$)
beta (β^-) decay	${}_{90}^{234}\text{Th} \rightarrow {}_{-1}^0e + {}_{91}^{234}\text{Pa}$	<ul style="list-style-type: none"> ● β^- particle = ${}_{-1}^0e$ (an electron) ● energy released in decay process <i>creates</i> the β^- particle (not from an orbital) ● net effect: convert neutron to proton ● β^- particles likely to be produced by nuclides with <i>high</i> neutron-to-proton ratios
positron (β^+) decay	${}_{19}^{38}\text{K} \rightarrow {}_1^0e + {}_{18}^{38}\text{Ar}$	<ul style="list-style-type: none"> ● β^+ particle = ${}_1^0e$ (a positive electron) ● β^+ is "antiparticle" of β^- ● net effect: convert proton to neutron ● β^+ particles likely to be produced by nuclides with <i>low</i> neutron-to-proton ratios
electron capture	${}_{80}^{195}\text{Hg} + {}_{-1}^0e \rightarrow {}_{79}^{195}\text{Au}$	<ul style="list-style-type: none"> ● an inner-orbital electron captured by nucleus ● generally slow ● net effect: convert proton to neutron ● likely for nuclides with <i>low</i> neutron-to-proton ratios
gamma (γ) decay	${}_{27}^{60}\text{Co} \rightarrow {}_{28}^{60}\text{Ni} + {}_{-1}^0e + 2 {}_0^0\gamma$	<ul style="list-style-type: none"> ● ${}_0^0\gamma$ = high energy photon ● frequently accompanies other decay types ● a way of "draining off" excess energy (product nuclide may be in excited state)
spontaneous fission	${}_{98}^{252}\text{Cf} \rightarrow {}_{54}^{140}\text{Xe} + {}_{44}^{108}\text{Ru} + 4 {}_0^1n$	<ul style="list-style-type: none"> ● generally slow ● "splitting" of heavy nuclide to lighter ones with similar mass numbers